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| FORM PTO-1390 DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE (REV 1-98) | | ATTORNEY'S DOCKET NO. MIDR695 |
| TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. 371 | | U.S. APPLICATION NO. (If known, see 37 CFR 1.5) 09/762035 |
| | | |
| INTERNATIONAL APPLICATION NO. PCT/EP99/05329 | INTERNATIONAL FILING DATE 16 JUL 1999 (16.07.1999) | PRIORITY DATE CLAIMED 30 JUL 1998 (30.07.1998) |
| TITLE OF INVENTION WELLBORE FLUID | | |
| APPLICANT(S) FOR DO/EO/US SAWDON, Christopher; BALLARD, David | | |
| Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information: | | |
| <ol style="list-style-type: none"> <input checked="" type="checkbox"/> This is a FIRST submission of items concerning a filing under 35 U.S.C. 371. <input type="checkbox"/> This is a SECOND or SUBSEQUENT submission of items concerning a filing under 35 U.S.C. 371. <input type="checkbox"/> This express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(1). <input checked="" type="checkbox"/> A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date. <input checked="" type="checkbox"/> A copy of the International Application as filed (35 U.S.C. 371(c)(2)). <ol style="list-style-type: none"> <input checked="" type="checkbox"/> is transmitted herewith (required only if not transmitted by the International Bureau). <input checked="" type="checkbox"/> has been transmitted by the International Bureau. <input type="checkbox"/> is not required, as the application was filed in the United States Receiving Office (RO/US). <input type="checkbox"/> A translation of the International Application into English (35 U.S.C. 371(c)(2)). <input checked="" type="checkbox"/> Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3)). <ol style="list-style-type: none"> <input type="checkbox"/> are transmitted herewith (required only if not transmitted by the International Bureau). <input type="checkbox"/> have been transmitted by the International Bureau. <input type="checkbox"/> have not been made; however, the time limit for making such amendments has NOT expired. <input checked="" type="checkbox"/> have not been made and will not be made. <input type="checkbox"/> A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)). <input type="checkbox"/> An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)). <input type="checkbox"/> A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)). | | |
| Items 11 to 16 below concern document(s) or information included: <ol style="list-style-type: none"> <input type="checkbox"/> An Information Disclosure Statement under 37 CFR 1.97 and 1.98. <input type="checkbox"/> An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included. <input type="checkbox"/> A FIRST preliminary amendment. <input type="checkbox"/> A SECOND or SUBSEQUENT preliminary amendment. <input type="checkbox"/> A substitute specification. <input type="checkbox"/> A change of power of attorney and/or address letter. <input checked="" type="checkbox"/> Other items or information: Return Postcard, Application Data Sheet, Petition for Revival | | |

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| CERTIFICATE OF EXPRESS MAILING | |
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| I hereby certify that this paper or fee is being deposited with the United States Postal Service "EXPRESS MAIL POST OFFICE TO ADDRESSEE" service under 37 C.F.R. 1.10 on the date indicated above and is addressed to Assistant Commissioner for Patents, Washington, DC 20231. | |
| <p align="center"><i>Carsten White</i></p> <p align="center">_____ Signature</p> | |

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|--|--------------|---|------------|---|--------|
| U.S. APPLICATION NO. (If known, see 37 CFR 1.5) 097/762635 | | INTERNATIONAL APPLICATION NO. PCT/EP99/05329 | | ATTORNEY'S DOCKET NUMBER MIDR695 | |
| 17. <input checked="" type="checkbox"/> The following fees are submitted: Basic National Fee (37 CFR 1.492(a)(1)-(5)): Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO \$1000.00 International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO \$860.00 International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO \$710.00 international preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4) \$690.00 International preliminary examination fee (37 CFR 1.482) paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) \$ 100.00 <div style="text-align: right;">ENTER APPROPRIATE BASIC FEE AMOUNT = \$1000.00</div> | | | | CALCULATIONS PTO USE ONLY | |
| | | | | | |
| Surcharge of \$130.00 for furnishing the oath or declaration later than <input type="checkbox"/> 20 <input checked="" type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492(e)). | | | | \$130.00 | |
| Claims | Number Filed | Number Extra | Rate | | |
| Total Claims | 13 - 20 = | 0 | x \$ 18.00 | \$0.00 | |
| Independent Claims | 3 - 3 = | 0 | x \$ 80.00 | \$0.00 | |
| Multiple dependent claim(s) (if applicable) | | | + \$270.00 | \$270.00 | |
| TOTAL OF ABOVE CALCULATIONS = | | | | \$1868.00 | |
| Reduction by 1/2 for filing by small entity, if applicable. Verified Small Entity statement must also be filed. (NOTE: 37 CFR 1.9, 1.27, 1.28) | | | | \$.00 | |
| SUBTOTAL = | | | | \$1868.00 | |
| Processing fee of \$130.00 for furnishing the English translation later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492(f)). | | | | \$n/a.00 | |
| TOTAL NATIONAL FEE = | | | | \$1868.00 | |
| Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property) | | | | \$.00 | |
| TOTAL FEES ENCLOSED = | | | | \$1868.00 | |
| | | | | Amount to be refunded: | \$.00 |
| | | | | charged | \$.00 |
| a. <input type="checkbox"/> A check in the amount of \$.00 cover the above fees is enclosed. b. <input checked="" type="checkbox"/> Please charge my Deposit Account No. <u>01-2508/MIDR695</u> in the amount of \$ <u>1868</u> to cover the above fees. A duplicate copy of this sheet is enclosed. c. <input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. <u>01-2508/MIDR695</u> . A duplicate copy of this sheet is enclosed. NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status. | | | | | |
| SEND ALL CORRESPONDENCE TO: Carter J. White HOWREY SIMON ARNOLD & WHITE LLP 750 Bering Drive Houston, TX 77057-2198 (713) 787-1400 | | | | <div style="text-align: center;">  SIGNATURE </div> <div style="text-align: center;"> Carter J. White NAME </div> <div style="text-align: center;"> <u>41,374</u> REGISTRATION NUMBER </div> | |

Wellbore Fluid

This invention relates to wellbore fluids suitable for use in oil and gas exploration and production industries and embraces fluids used for drilling, under-reaming, completion, cementing, fracturing, stimulation, workover and packing of wellbores and also includes spacer fluids whose function is to separate two fluids during pumping operations and spotting fluids whose function is to treat certain intervals of the wellbore.

In the process of rotary drilling a well, a drilling fluid or mud is being circulated down the rotating drill pipe, through the bit, and up the annular space between the pipe and the formation or steel casing, to the surface. The drilling fluid performs different functions such as removal of drilled cuttings from the bottom of the hole to the surface, suspension of cuttings and weighting material when circulation is interrupted, control of subsurface pressures, maintaining the integrity of the wellbore until the well section is cased and cemented, isolate the fluids from the formation by providing sufficient filtration control to prevent excessive loss of fluids to the formation, cool and lubricate the drill string and bit, maximise penetration rate etc.

The required functions can be achieved by a wide range of fluids composed of various combination of solids, liquids and gases and classified according to the constitution of the continuous phase mainly in two groupings : aqueous (water-based) drilling fluids, and non-aqueous (mineral oil or synthetic-base) drilling fluids, commonly 'oil-base fluids'

Other types of fluid used in well operations include completion fluids, a term which commonly refers to fluids pumped after drilling finishes but prior to starting production, and workover fluids, used in remedial operations usually on a well that is already producing.

During the operations for drilling and completing hydrocarbon-bearing formations (reservoirs), an overbalance pressure is often applied which causes fluid loss from the wellbore into the reservoir rock. This filtration process causes solid particles to block pores in the formation with the build-up of a low permeability internal filtercake comprised of the solid phases present in the fluid as shown figure 1. The depth of invasion may be from a few millimetres to (occasionally) many centimeters. This is the primary cause of damage (loss of productivity) in open hole wells, a phenomenon which is increased by a large overbalance pressure. Furthermore the formation of a thick filter cake may

lead to an increased risk of getting drill pipes or measuring tools stuck in the wellbore and to failure in cementing the wellbore casings.

Mud solid invasion is also important when the well comprises large natural fractures as shown figure 2. The fractures invaded by the mud are no longer available to drain the oil into the wellbore.

5 In workover operations, perforations may be also invaded in the same way.

It is therefore highly desirable to provide means to at least minimise formation damage.

High fluid loss, especially spurt loss, tends to increase mud invasion. Accordingly, additives of a polymeric type such as biopolymers (Xanthan, Scleroglucan), starches and celluloses (hydroxyethyl cellulose[HEC], polyanionic cellulose [PAC]) - are added to provide viscosity and fluid loss control.

10 Usually wellbore fluids also contain inorganic solids such as clays, barite and calcium carbonate. When considering minimising mud invasion, an important aspect in selecting such inorganic solids is their particle size distribution for the particles to seal the entrance to pores or fractures in the reservoir rock. The "bridging" solids are combined with water soluble or colloidal polymers to enhance the seal.

15 Sized particles limit the depth of invasion as shown figure 3 and 4 where the pore or fracture entrance is sealed with bridging solids. The solids used include for instance ground silica (especially common in fracturing operations), mica, calcium carbonate, ground salts and oil soluble resins.

20 Another often interrelated approach is to use soluble solids to allow subsequent clean-up with wash fluids. Accordingly sized calcium carbonate (that can be dissolved by acid and therefore be removed from the pores) is a typical component of drill-in fluids. The polymers used in conjunction with the calcium carbonate are often selected on the basis that they can also be broken down by acid or enzyme treatments to prevent them impairing the permeability of the formation.

25 However, these remedial treatments involving the use of strong acid (for instance 15% HCl) have a high cost and can be indeed hazardous and ineffective for the following reasons. Poor dissolution of the filter cake can be caused by zones of higher permeability channelling acid away and into the formation. This can lead to further formation damage. Strong acids involve health and safety issues and cause corrosion of sand screens and downhole equipment. Treating a wellbore interval also raises placement issues. Moreover, laboratory tests have shown that the acids can be damaging to the reservoir rock matrix.

Because of these limitations there are many systems and products on the market to try and improve the performance of filter cake clean up treatments. However, these to have associated problems. For example, the use of alkaline oxidising agents can lead to problems of iron oxide scale in metal tubulars. Enzymes take a long time to react and are limited in application by temperature, pH, and salinity. Internal breakers, such as magnesium peroxide, can be added to the reservoir drill-in fluid so it forms part of the filter cake. This is then activated by acid to produce hydrogen peroxide to facilitate the breakdown the polymers in the cake. Unfortunately, this can actually happen in the reservoir drill in fluid whilst drilling which can have a negative effect on fluid properties.

The present invention aims at providing new wellbore fluids which can form an easy-to-remove filter cake.

The invention provides a fluid system that, when used for the purposes of drilling, completing, cementing, stimulating, packing or working over a wellbore, will facilitate filter cake removal. These can be aqueous or non-aqueous based fluids (such as hydrocarbon based fluids). The result will be reduced or often negligible impairment of permeability of the producing formation by the relevant reservoir servicing fluid compared with well servicing fluids containing conventional bridging materials.

In accordance with the present invention, acetal crosslinked polymers are used as the bridging agent in the relevant fluid. In essence the bridging material is made of a particulate material which may swell but is substantially insoluble in water under non-acidic conditions (at pH >7.5) and which is degraded at acidic pH (below 6.5) to substantially solids-free or soluble decomposition products. The particles are composed of the reaction product of A) one or more water soluble organic compound having possessing a molecular weight of less than 30,000 and possessing at least two hydroxyl groups and B) any other organic compound(s) capable of forming acetal or hemiacetal cross-links with the hydroxyl groups of compound A.

Examples of compounds A with free hydroxyl groups capable of entering into the crosslink reaction of the invention include monosaccharides, oligosaccharides, polysaccharides of molecular weight less than 30,000, glycerol, polyglycerols, erythritol, pentaerythritol, mannitol, sorbitol, glycols, polyalkylene glycols, and low molecular weight water soluble vinyl polymers possessing hydroxyl groups.

Suitable acetal crosslink agents B include aliphatic monoaldehydes and dialdehydes having from 2 to 10 carbon atoms and esters of propiolic acid wherein the alcohol forming the ester has from 1 to 8 carbons. A preferred compound is 1,5 pentanedial (glutaric dialdehyde) $C_5H_8O_2$.

Usually, the amount of cross-linking agent varies from 0.5 to 15% w/w (and the amount of compound A accordingly varies from 99.5 to 85% w/w).

An example of a useful material has been prepared from 95% yellow dextrin (a gum produced from the acid hydrolysis and depolymerisation of starch feedstock) and 5% by weight pentanedial.

Acetal crosslinked starches are known from U.S. Patents 4,048,435 and 4,098,997 that describe methods of preparing reversibly crosslinked granular starches for use in the paper, adhesives and textile industries but with no mention as to their potential use in the oil industry. It is stated that these starches contain acetal crosslinkages, which are stable under neutral or alkaline conditions but which readily hydrolyse under acidic conditions, and they are designed to dissolve and form a colloidal suspension in water. By contrast, the present invention contemplates the cross-linking of much lower molecular weight polyols than the above mentioned starches to produce a particulate suspension where the particles do not dissolve or disperse at alkaline pH.

The solid may be ground or prepared to any desired particle size.

The particles are stable under alkaline conditions and insoluble (but usually swell) in aqueous wellbore fluids of pH 7.5 and above. Thus they provide very effective sealing of the formation by virtue of the swollen particles' ability to deform to "fit" the pore openings or fractures.

Most advantageously however, the cross links rapidly hydrolyse under acidic catalysed conditions, under even weakly acidic conditions at pH 6.5 and below. The degradation products are essentially solids free water soluble moieties of low molecular weight and low viscosity in solution. Hence any flow restriction (of hydrocarbons) caused by the particles may be efficiently removed. A low pH solution such as a weak easily-handled acid of little corrosion potential may be pumped in place to catalytically degrade the particles. Alternatively, produced fluids, be they oil and water or gas and water, usually exhibit an aqueous pH of less than pH 6.5 due to the carbon dioxide commonly present. Hence the removal of the temporary seal may be triggered simply by allowing the well to flow. As the seal degrades the well flow increases autocatalytically.

The hydrolysis does not consume the acid. Hence much enhanced clean up and optimised well productivity is anticipated compared to that exhibited when using current materials such as ground calcium carbonate.

The following features and benefits have been identified with the use of acetal crosslinked, polymeric, bridging solids:

- Filter cakes made from these novel materials almost totally degrade in the presence of weak acid giving almost 4x the return flow rate compared to a corresponding calcium carbonate based fluid under the same conditions. This also gives the benefits of better health and security from handling weak acids and less corrosion of downhole screens and equipment.
- A feature of the invention is that the degradation of the acetal bridging material is acid catalysed, as opposed to the conventional stoichiometric reaction. This means the acid is not consumed or neutralised in the reaction. The benefit of this is an enhanced acid treatment as the acid is available to go on degrading more filter cake.
- Due to the high mobility of the hydrogen ion, (which can therefore penetrate the water-swollen particles) the acid does not have to be forced into the filter cake to break it down. Which means that less acid induced damage results from strong acid being forced into the formation causing fines to migrate, bridge and restrict production.
- Acetal crosslinked bridging materials can degrade under simulated well flowing conditions in the laboratory, where the weak acid is provided by produced CO₂ forming carbonic acid in the aqueous phase. This could negate the need for a costly wash fluid treatment.
- Acetal bridging materials have a low density. A low density will enable a wider range of applications to be covered, density constraints often limit the amount of bridging solid which can be used. Calcium carbonate has a density of 2.7 S.G. and cannot always be added at the desired concentrations for bridging in low density fluids. Another benefit will be improved removal of undesirable, non acidisable, contaminating solids, from the fluid. Contaminating solids from the formation have roughly the same density as calcium carbonate which makes separation difficult in a conventional fluid. However, these solids are approximately 2.7 times denser than the acetal bridging solid. This means that highly efficient separation techniques, such as centrifugation, can be employed to remove non acidisable solids from the fluid without removing the acetal bridging solids. This therefore makes the fluid less formation damaging.

Preparation of an acetal bridging solid.

95 weight parts of dry yellow dextrin was dissolved in water and 5 weight parts pentanediol added. The sample was mixed and heated to 130°C. The water was then removed under partial vacuum over a 16 hour period. The solid from this was then ground to a fine particle size (substantially all less than 150 microns) and tested as follows.

The wellbore fluids in the following examples have been tested using a high pressure/high temperature fluid loss cell as schematically represented figure 5. Such an apparatus typically comprises a measuring cell 1 closed by an upper cover provided with a central passage 2 and a bottom provided with a drain 3. The bottom is covered with a base made of a porous material, in this instance a 5 micron aloxite disc. Gas can be injected to provide the differential pressure between the mud and the formation fluids. The product doses are expressed in pounds per barrel (ppb).

Example 1

A fluid system of this invention (Acetal fluid) was compared to one of conventional formulation, CaCO₃ based fluid (comparison fluid) containing 49ppb calcium carbonate bridging solids, 6ppb DUALFLO filtration control additive, 1ppb IDVIS viscosifier and suspending agent, 25ppb CaCl₂, 2ppb PTS200 alkaline buffer.

The Acetal fluid contains - 40ppb ACETAL bridging solids, 6ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 2ppb PTS200.

The testing of the fluids was according to the following protocol, schematically illustrated figure 5 where the arrows indicate the direction of the fluid flow in the core plugs (the upwards arrow corresponds to fluid loss from the subterranean rock to the wellbore while the downwards arrow characterises fluid loss from the wellbore into the subterranean rock).

- Step 1: 'apply' mud damage, using the tested wellbore fluid, to a 5 micron aloxite disc under typical well conditions (temperature 80° C, pressure of 500 psi). Fluid loss is measured after 15.5 hours (downwards arrow). A filter cake is formed on the aloxite disc.
- Step 2: cell emptied hot and refilled with 10% NaCl brine and 12/20 sand. A 45 micron aperture screen was then placed on the sand. The sand is used for modelling a gravel pack which may be placed in a well to control sand production.
- Step 3: simulates the flow from the reservoir oil into the wellbore through the filter cake (upwards arrow) , using kerosene flow back and a typical formation pressure of 4 psi with carbon dioxide as the pressuring gas at 80°C.
- Step 4 : Flow 500g of CO₂ containing kerosene, shut the cell and leave for 64 hours at 80°C and then test the flow rate in the production direction

| Sample | 15.5 hr. Fluid loss | Flow rate after shut in |
|-----------------------------------|---------------------|-------------------------|
| | (ml) | (g/min) |
| CaCO ₃ bridging solids | 17 | 22..2 |
| Acetal bridging solids | 20 | 33.6 |

The example illustrates the effect that an ACETAL fluid filter cake is degraded in the presence of kerosene containing dissolved carbon dioxide. The return flow rate for the ACETAL fluid was 1.5 x higher than for the CaCO₃ fluid. This test suggests that naturally occurring fluids from a production zone will degrade the filter cake even in the absence of applied acid. It can also be seen that using all acetal bridging solids gives a slightly higher fluid loss. It was found that the use of a small amount of calcium carbonate in conjunction with the acetal solids reduced the filtrate loss, allowing a lower concentration of DUALFLO fluid loss additive to be utilised in the following examples.

Example 2

The CaCO_3 based fluid contains - 40ppb calcium carbonate, 6ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 1ppb PTS100.

The ACETAL based fluid contains - 7ppb calcium carbonate, 30ppb ACETAL solids, 3ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 3ppb PTS100.

The following test protocol is used

- Step 1: 'apply' mud damage, using the tested wellbore fluid, through a 5 micron aloxite disc at a temperature of 80° C and a pressure of 500 psi. The fluid loss is measured after 15.5 hours.
- Step 2: cell emptied hot and refilled with 100 ml 5% citric acid as a moderated acidic wash fluid.
- Step 3: Reseal cell, do not pressurise, leave 72 hours at 80°C
- Step 4 : Open cell and fill with 12-20 sand and fit 45 micron aperture screen
- Step 5: Carry out kerosene return flows at 4psi, at a temperature of 80°C.

| Sample | 16.5 hr. Fluid loss (ml) | Flow rate (g/min) | pH of produced fluids |
|-----------------------|-----------------------------|-------------------|--------------------------|
| CaCO_3 fluid | 23 | 21.2 | 5.9 |
| ACETAL fluid | 17 | 80.9 | 2.0 |

The example signifies that under mild acid conditions the return flow rate for the ACETAL fluid was nearly 4x higher than for the CaCO_3 fluid, indicating that the ACETAL fluid filter cake has degraded to a much greater extent. This was confirmed on opening the cell which showed the presence of a large filter cake for the CaCO_3 fluid and little to no filter cake for the ACETAL fluid.

The lower pH of the produced fluids from the cell containing the ACETAL filter cake indicates that the acid is not being used up to the same extent as with CaCO_3 filter cake. The example also

demonstrates that the filter cake is degraded without pressurisation of the cell. Conventionally, these tests are typically carried out under pressure to force acid into the cake.

Example 3

- 5 The CaCO_3 based fluid contains - 40ppb calcium carbonate, 6ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 2ppb PTS200, 15ppb OCMA clay.

The ACETAL based fluid contains - 7ppb calcium carbonate, 30ppb ACETAL solids, 3ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 3ppb PTS100 15ppb OCMA clay.

- 10 The test protocol of example 2 was used except that step 1 lasted only 15.5 hours and step 3 20hours.

| Sample | 15.5 hr. Fluid loss | Flow rate |
|---|---------------------|-----------|
| | (ml) | (g/min) |
| CaCO_3 fluid | 32 | 9.8 |
| ACETAL fluid | 18.8 | 21.4 |

- 15 The example demonstrates that an ACETAL fluid filter cake containing non acidisable OCMA clay is degraded by weak acid in a short time. The return flow rate for the ACETAL fluid was 2.2 x higher than for the CaCO_3 fluid, indicating that the ACETAL fluid filter cake has degraded to a much greater extent.

Example 4

- 20 The CaCO_3 based fluid contains - 40ppb calcium carbonate, 6ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 2ppb PTS200.

The ACETAL based fluid contains - 10ppb calcium carbonate, 30ppb ACETAL solids, 3ppb DUALFLO, 1ppb IDVIS, 43ppb KCl, 2ppb PTS200.

The following test protocol was used :

- 5 • Step 1: 'apply' mud damage, using the tested wellbore fluid, through a 5 micron aloxite disc at a temperature of 80° C and a pressure of 500 psi. The fluid loss is measured after 15.5 hours.
- Step 2: cell emptied hot and refilled with 0.1m (0.37%w/v) HCl and 12/20 sand and a 45 micron aperture screen fitted.
- Step 3: The cell was resealed and left for 4hours @ 80°C
- 10 • Step 4: then kerosene flow backs carried out at 4psi and 80°C.

| Sample | 15.5 hr. Fluid loss (ml) | Flow rate (g/min) |
|-------------------------|--------------------------|-------------------|
| CaCO ₃ fluid | 29 | 20.7 |
| ACETAL fluid | 16 | 32.0 |

The example demonstrates that exposure of ACETAL fluid filter cake to even a highly diluted acid causes it to degrade in only a short time. The return flow rate for the ACETAL fluid was 1.5 x
15 higher than for the CaCO₃ fluid.

Claims

1. A wellbore fluid comprising a water-non soluble particulate material composed of the reaction product of A) one or more water soluble organic compound having possessing a molecular weight of less than 30,000 and possessing at least two hydroxyl groups and B) any other organic compound(s) capable of forming acetal or hemiacetal cross-links with the hydroxyl groups of compound A.
2. The wellbore fluid of claim 1 wherein hydroxyl compound (A) is selected from the class including monosaccharides, oligosaccharides, polysaccharides of molecular weight less than 30,000, glycerol, polyglycerols, erythritol, pentaerythritol, mannitol, sorbitol, glycols, polyalkylene glycols, and low molecular weight water soluble vinyl polymers possessing hydroxyl groups.
3. The wellbore fluid according to claim 1 or 2, wherein compound (B) is selected from the class including aliphatic aldehydes and dialdehydes having from 2 to 10 carbon atoms, and esters of propiolic acid wherein the alcohol forming the ester has from 1 to 8 carbon atoms.
4. The wellbore fluid according to any of the preceding claim, wherein 0.5-15%, dry weight of compound (A) and 95.5-85% of said compound (B) is reacted.
5. The wellbore fluid according to any of the preceding claim wherein the particulate material is the reaction product of dextrin and pentanedial.
6. Application of the wellbore fluid according to any of the preceding claim as any of the following well processes : drilling, under-reaming, completing, working over, sealing loss zones, sealing fractures, sealing cavities or other very high permeability conduits in a rock formation, or hydraulic fracturing to stimulate a hydrocarbon-producing zone
7. A process wherein after any of the well processes of claim 6, a low pH fluid containing any acid or buffered solution of less than pH 6.0 is pumped into the producing zone segment of the wellbore to catalyse the decomposition of the particulate material of this invention.

-
- Figure 1 consists of 12 subplots, labeled (a) through (l), showing the evolution of the electron distribution function $f(v)$ versus velocity v . The x-axis for all plots is velocity v , ranging from -10 to 10. The y-axis is the distribution function $f(v)$, ranging from 0 to 1.0. The plots show the development of a distribution from an initial state (a) to a state with a sharp peak at $v=0$ (l). The subplots are arranged in a 6x2 grid. The first column (a-f) shows the distribution at times $t = 0, 1, 2, 3, 4, 5$. The second column (g-l) shows the distribution at times $t = 6, 7, 8, 9, 10, 11$. The distribution starts as a broad, flat curve at $t=0$ and evolves into a sharp peak at $v=0$ by $t=11$.

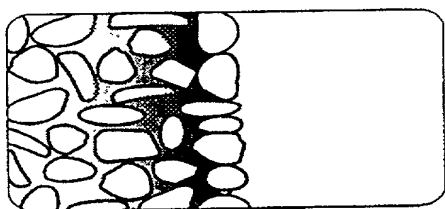


Figure 1

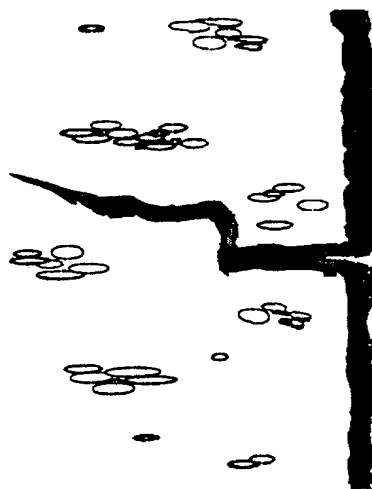


Figure 2

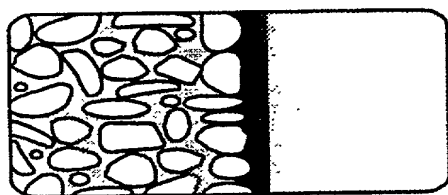


Figure 3



Figure 4

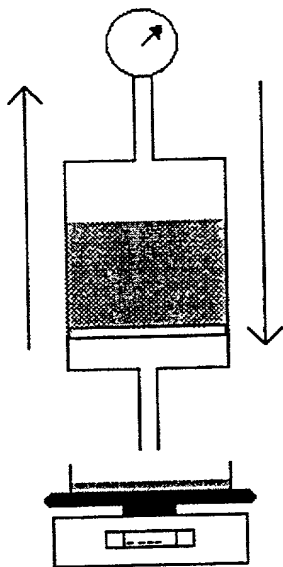


Figure 5

**DECLARATION FOR PATENT APPLICATION AND
POWER OF ATTORNEY**

As a below named inventor, I hereby declare that:

My residence, Mailing address and citizenship are as stated below next to my name,

I (we) believe we are the original, first and joint inventor of the subject matter which is claimed and for which a patent is sought on the invention entitled

WELLBORE FLUID

the specification of which

☒ is attached hereto.

☒ was internationally filed on July 19, 1999 as
International PCT/EP99/05329
Application Serial No. _____
and was amended _____
(if applicable)

I (we) hereby state that we have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I (we) acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

☒ I (we) hereby claim foreign priority benefits under Title 35, United States Code, §119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Prior Foreign Application(s):

| | | | <u>Priority Claimed</u> | |
|-----------------------|-----------|------------------------|-------------------------------------|--------------------------|
| <u>PCT/EP99/05329</u> | <u>GB</u> | <u>July 19, 1999</u> | <input checked="" type="checkbox"/> | <input type="checkbox"/> |
| (Number) | (Country) | (Day/Month/Year Filed) | Yes | No |
| <u>9816479.1</u> | <u>GB</u> | <u>July 30, 1998</u> | <input checked="" type="checkbox"/> | <input type="checkbox"/> |
| (Number) | (Country) | (Day/Month/Year Filed) | Yes | No |

☒ I (we) hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the

first paragraph of Title 35, United States Code, §112, I (we) acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

| | | |
|--------------------------|---------------|--------------------------------|
| PCT/EP99/05329 | July 16, 1999 | Pending |
| (Application Serial No.) | (Filing Date) | (Status) |
| | | (Patented, Pending, Abandoned) |
| (Application Serial No.) | (Filing Date) | (Status) |
| | | (Patented, Pending, Abandoned) |
| (Application Serial No.) | (Filing Date) | (Status) |
| | | (Patented, Pending, Abandoned) |
| (Application Serial No.) | (Filing Date) | (Status) |
| | | (Patented, Pending, Abandoned) |

I (we) hereby appoint the following as our representative(s) to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith: **STEPHEN H. CAGLE**, Attorney (Reg. No. 26,445), **PATRICIA A. KAMMERER**, Attorney (Reg. No. 29,775), **CRAIG M. LUNDELL**, Attorney (Reg. No. 30,284), **JANELLE D. WAACK**, Attorney (Reg. No. 36,300), **CARTER J. WHITE**, Agent (Reg. No. 41,374), and **ROBERT P. AUERBACH**, Agent (Reg. No. 46,525) each an attorney or agent with the law firm of HOWREY, SIMON, ARNOLD, & WHITE, LLP as its attorney or agent so long as they remain with such law firm.

Address all telephone calls to:
at telephone No. (713) 268-1372

Carter J. White



Address all correspondence to:

Customer No.

27551
PATENT TRADEMARK OFFICE

HOWREY, SIMON, ARNOLD, & WHITE
750 Bering Dr.
Houston, TX 77057

I (we) hereby declare that all statements made herein of our own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under §1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

Full name of first inventor Christopher Alan Sawdon

First Inventor's signature C Sawdon 10th May 2001 Date

Residence 6, Southview Road, Biscovey, Par, Cornwall, PL24 2HJ United Kingdom GBX

Citizenship United Kingdom

Mailing Address Same as above

Full name of second joint inventor, if any: David Antony Ballard

Second Inventor's signature _____ Date

Residence 6, William Mackie Ave., Stonehaven, Scotland, AB39 2PQ United Kingdom

Citizenship United Kingdom

Mailing Address Same as above

Full name of third joint inventor, if any _____

Third Inventor's signature _____ Date

Residence _____

Citizenship _____

Mailing Address _____

Full name of first inventor Christopher Alan Sawdon

First Inventor's signature _____

Date _____

Residence 6, Southview Road, Biscovey, Par, Cornwall, PL24 2HJ United Kingdom

Citizenship United Kingdom

Mailing Address Same as above

Full name of second joint inventor, if any: David Antony Ballard

Second Inventor's signature D Ballard 10/05/01

10/05/01

Date _____

Residence 6, William Mackie Ave. Stonehaven, Scotland, AB39 2PQ United Kingdom

Citizenship United Kingdom

Mailing Address Same as above

Full name of third joint inventor, if any

Third Inventor's signature _____

Date _____

Residence _____

Citizenship

Mailing Address _____